

---

# SunLab Molten-Salt Thermal Storage Tests

James E. Pacheco

## Team Members:

Mary Jane Hale, Hank Price, Bill Kolb, Vahab Hassani,  
Dan Blake, Steve Showalter, Kye Chrisman, Jim  
Grossman, Darrell Johnson, David Johnson, JJ Kelton,  
Earl Rush, Steve St. Laurent



Concentrating Solar Power

## Sun♦Lab

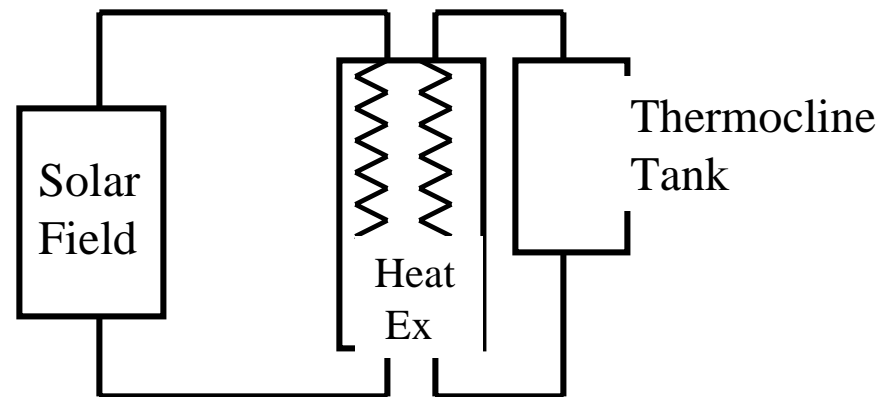
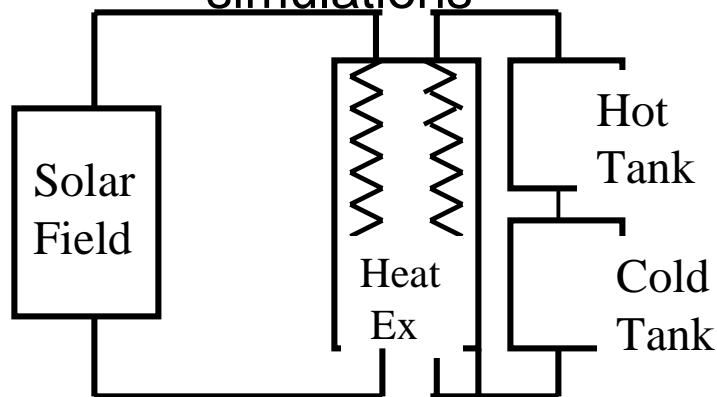
Sandia National Laboratories, Albuquerque, NM  
National Renewable Energy Laboratory, Golden CO

---

# Approach

- Near-Term Storage

- Emphasis on indirect, molten-salt systems (two tank and thermocline)
- Development and Testing: prototype at Sandia
- Analysis: finite difference and CFD for thermocline, annual simulations



Concentrating Solar Power

## SunLab

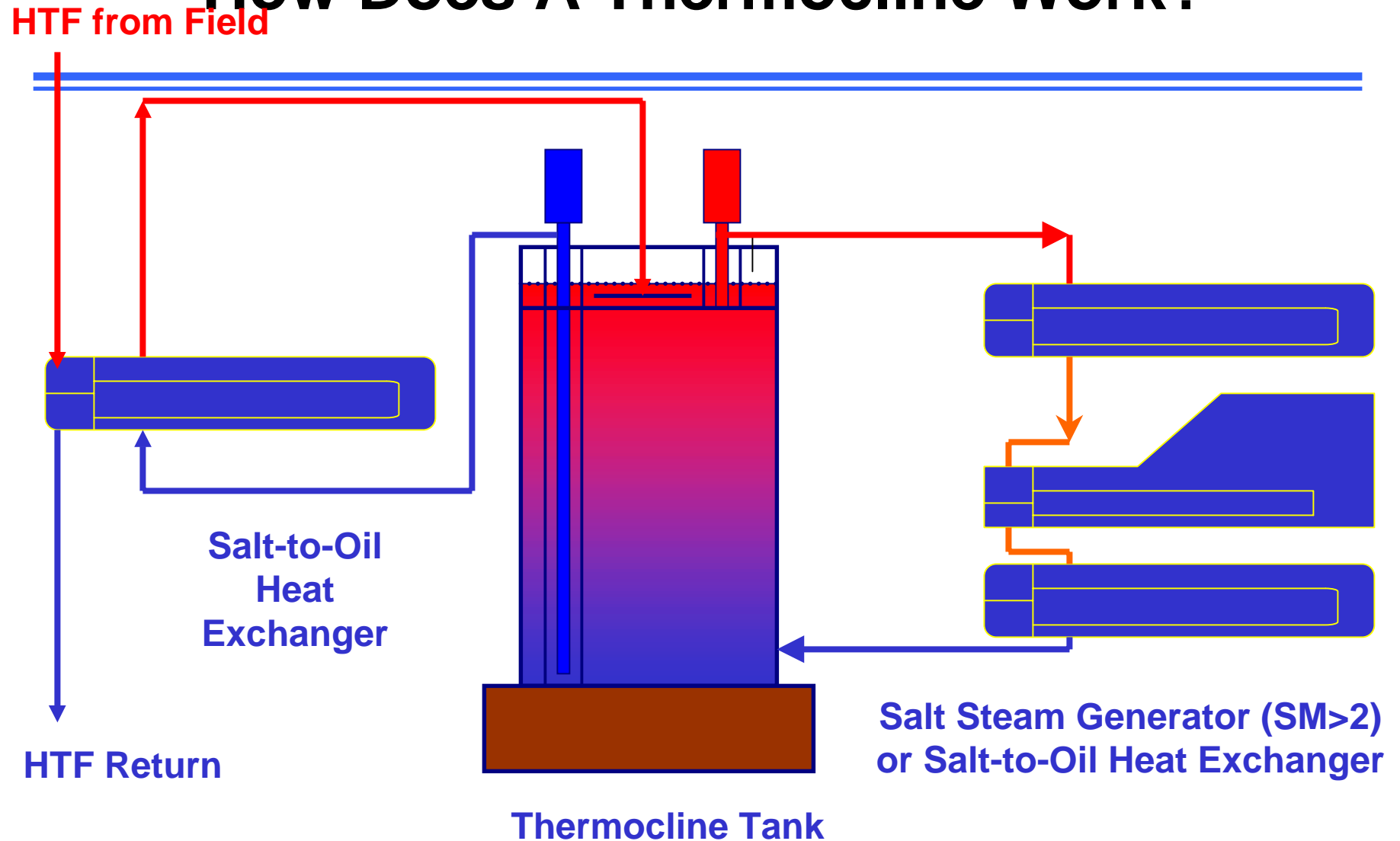
Sandia National Laboratories, Albuquerque, NM  
National Renewable Energy Laboratory, Golden CO

# Why Are We Interested in Molten Salt Thermocline Systems?

---

- Potentially lower cost than other options (currently \$25-\$35/kWh)
- Simple system design
- Compact, more efficient
- Delivers a large portion of its energy at nearly constant outlet temperature
- Experience from Solar One's thermocline
- Experience with molten salt

# How Does A Thermocline Work?



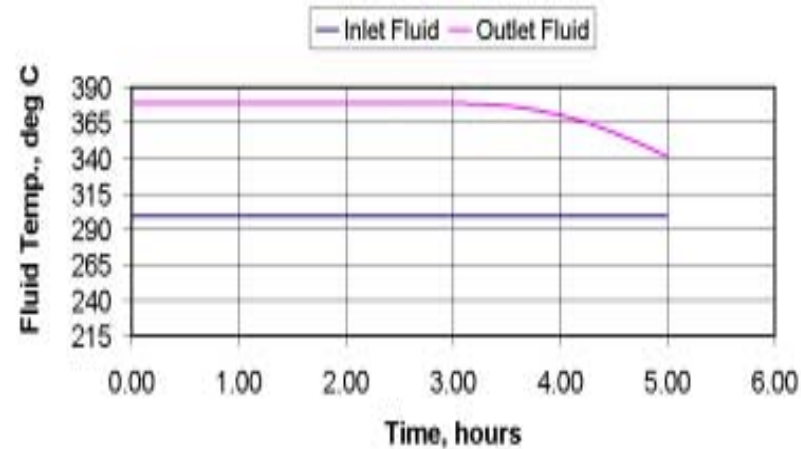
Concentrating Solar Power

## SunLab

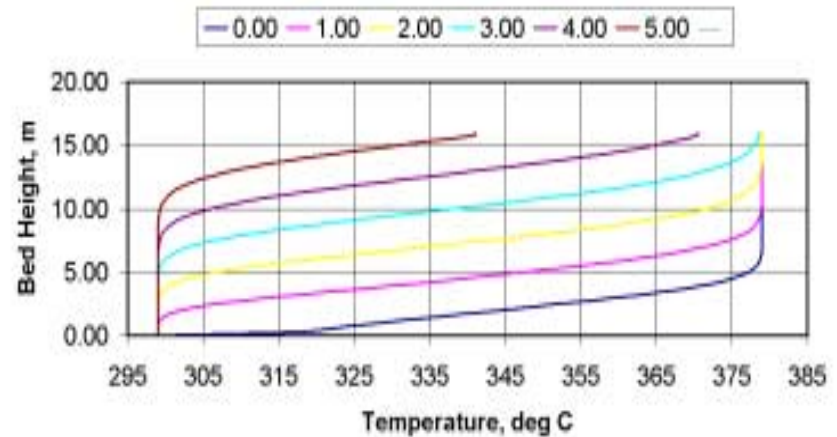
Sandia National Laboratories, Albuquerque, NM  
National Renewable Energy Laboratory, Golden CO

# Discharging and Charging

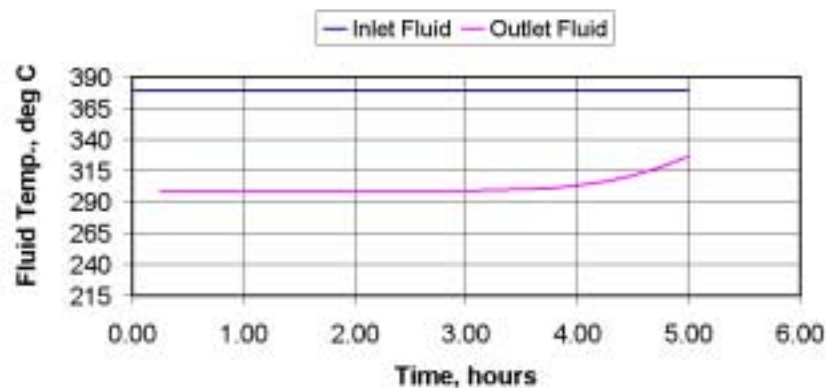
## Discharging



## Temperature Profiles



## Charging



# Candidate Filler Materials

Name	Formula
Alumina	$\text{Al}_2\text{O}_3$
Corundum*	$\text{Al}_2\text{O}_3$
Bauxite*	$\text{AlO}_x(\text{OH})_z$
Witherite*	$\text{BaCO}_3$
Barite*	$\text{BaSO}_4$
Marble*	$\text{CaCO}_3$
Fluorite*	$\text{CaF}_2$
Anhydrite*	$\text{CaSO}_4$
Taconite*	$\text{Fe}_2\text{O}_3, \text{Fe}_3\text{O}_4$
Ilmenite*	$\text{FeTiO}_3$
Magnesite*	$\text{MgCO}_3$
Silicon carbide	$\text{SiC}$
Cassiterite*	$\text{SnO}_2$
Hydroxyapatite	$\text{Ca}_5(\text{PO}_4)_3(\text{OH}, \text{F}, \text{Cl})$
Fluorapatite	$\text{Ca}_5(\text{PO}_4)_3(\text{OH}, \text{F}, \text{Cl})$
Limestone	$\text{CaCO}_3$
Quartzite	$\text{SiO}_2$



Concentrating Solar Power

## SunLab

Sandia National Laboratories, Albuquerque, NM  
National Renewable Energy Laboratory, Golden CO

# What Are the Technical Risks

---

- Compatibility of filler materials with molten salt
- Unproven concept
- Difficulty working with nitrate salt (freezing)
- Danger of having a fuel (Therminol) next to an oxidizer (nitrate salts)

# Near-Term Thermal Storage Development Activities

---

- Isothermal compatibility tests
  - Objective: Evaluate potential filler materials
  - 17 candidate materials
  - Measure weight loss in nitrate salt for 1000 h
  - Analyze salt chemistry
  - Status: Complete 10h, 100h, and 1000h tests. Salt analysis partially completed.



Barite: Before and After 100 h

Witherite: Before and After 100 h

Taconite: Before and After 100 h

NM Limestone: Before and After 100 h

Quartzite: Before and After 100h



# Near-Term Thermal Storage Development Activities (Continued)

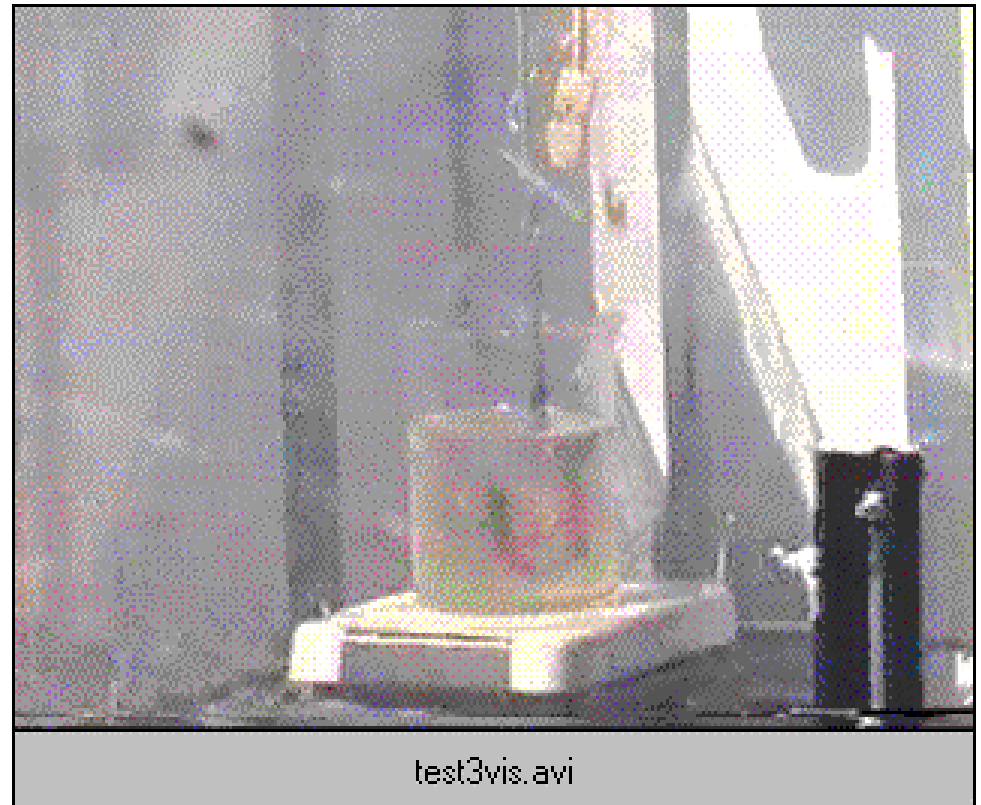
- Thermal Cycling Test
  - Objective: Evaluate mechanical integrity of best candidate filler materials
  - Cycle with flowing salt between 290 C and 400 C
  - Status: Completed 360 cycles on Taconite, Limestone and Marble, 550 cycles on Quartzite.
  - Limestone did not hold up to thermal cycling.
  - Marble shows possible recrystallization.
  - Taconite and Quartzite held up well. Currently testing Silica sands.



“Hot Rocks”  
Thermal Cycling System

# Near-Term Thermal Storage Development Activities (Continued)

- Salt Safety Tests:
  - Objective: Evaluate reaction of Therminol oil with nitrate salt in the event of a breach of the salt-to-oil heat exchanger
  - Introduced Therminol into beaker of nitrate salt at 400 C:
    - 1) on surface of salt, 2) below surface, 3) trapped in scintillation vial held below surface, and 4) spark above vapor tapped above salt
  - Results:
    - Therminol rapidly boiled off
    - No evidence of reaction with salt
    - Vapor ignited in test 4 (as



# Near-Term Thermal Storage Development Activities (continued)

- Engineering-Scale Thermocline Test
  - Objective: Verify operation of a small engineering-scale thermocline (capacity, operability, temperature gradient). Compare with modeled behavior.
  - 1.5 MWh capacity (3 m diameter x 6 m tall)
  - Heat input simulated with propane heater, heat rejection with air-blast cooler
  - Filler material and salt mixture TBD
  - Status: System designed and laid out. Tank fabricated and mounted. Heater being refurbished. Cooler mounted. Piping installed. Testing planned to begin this summer.



Concentrating Solar Power

## Sun Lab

Sandia National Laboratories, Albuquerque, NM  
National Renewable Energy Laboratory, Golden CO